

تم تحميل هذا الملف من موقع المناهج الإماراتية



## شرح درس solutions and Water والمحاليل منهج انسابير

موقع المناهج ← المناهج الإماراتية ← الصف العاشر المتقدم ← علوم ← الفصل الأول ← ملفات متنوعة ← الملف

تاريخ إضافة الملف على موقع المناهج: 15:10:10 2024-12-01

ملفات اكتب للمعلم اكتب للطالب | اختبارات الكترونية | اختبارات | حلول | عروض بوربوينت | أوراق عمل  
منهج انجليزي | ملخصات وتقارير | مذكرات وبنوك | الامتحان النهائي للمدرس

المزيد من مادة  
علوم:

إعداد: أحمد الحداد

## التواصل الاجتماعي بحسب الصف العاشر المتقدم



صفحة المناهج  
الإماراتية على  
فيسبوك

الرياضيات

اللغة الانجليزية

اللغة العربية

التربية الاسلامية

المواد على تلغرام

## المزيد من الملفات بحسب الصف العاشر المتقدم والمادة علوم في الفصل الأول

شرح درس الذرات والعناصر والمركبات منهج إنسابير

1

ملخص وشرح الدرس الثالث Compounds Carbon مكونات الكربون

2

ملخص وشرح الدرس الثاني Water of Properties خصائص الماء

3

ملخص وشرح الدرس الأول Reactions Chemicals التفاعلات الكيميائية

4

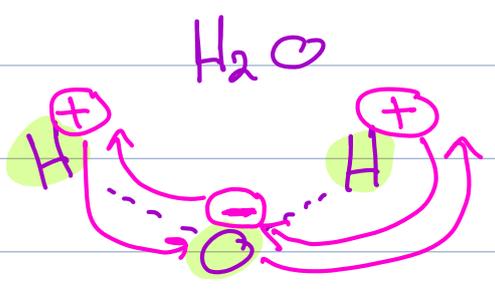
ملخص وشرح الدرس الأول System Circulatory الجهاز الدوري

5



Water and Solutions:-

Water (وجعلنا من الماء كل شيء حي)



- 70% of earth water
- 70% of organisms water
- 70% of cell water

water is polar molecule

قطبى has - and + charge.

ميزتاسي      O      H<sub>2</sub>      شحرات متعاكسة

water unequal distribution of charges. (opposite charges)

Because the water has opposite charges, it conduct heat and electric.

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كهرباء ساكنة

② Cause electrostatic

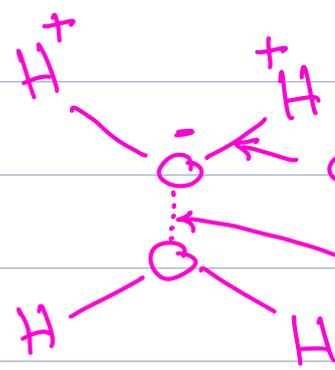
attraction between molecules. (van der waals forces)

تطفو

قوة فاندرفال

- allow leaves to float.
- allow some organisms to walk above water (gekkidae)

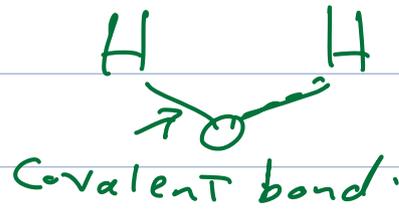
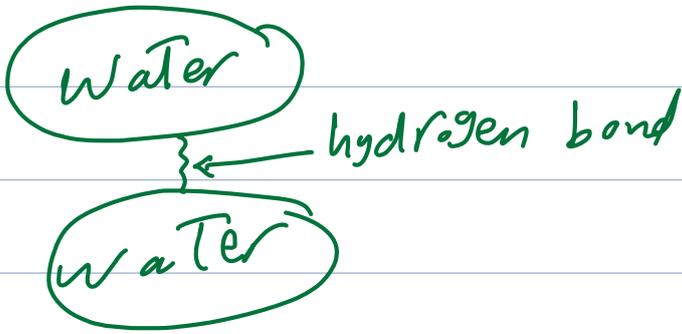
between O + H in water.



covalent bond strong / share electrons.

hydrogen bond weak between 2 water molecule

رابطة هيدروجينية ضعيفة



\* Capillary action: **الخاصية الشعرية**

allow water to move from roots to stems and leaves through Xylem Tissue.

**نسيج الخشب**

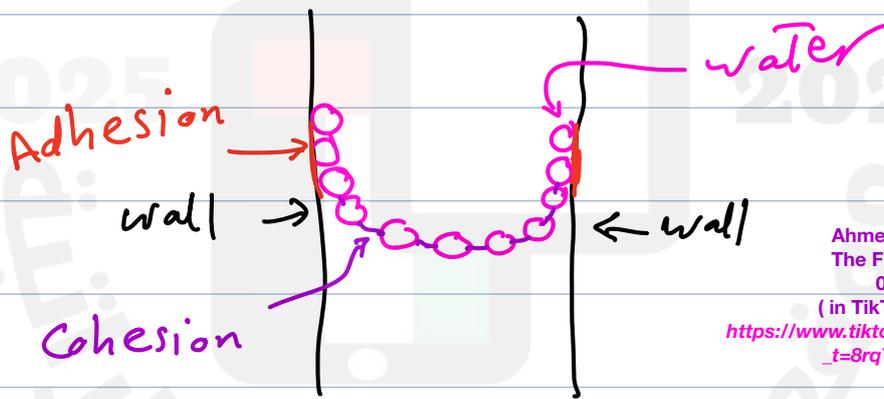


**Cohesion**

Bond between water and water. (same substance)

**Adhesion**

Bond between water and walls (Xylem).



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\* Cohesion and adhesion **سبب** Cause water allow **strider** and leaves stay float.

**متزليج بالي**

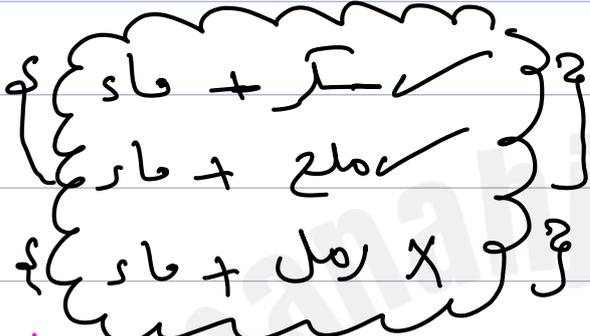
Surface Tension  
**التوتر السطحي**

# \* Properties of water:



- ① water is polar molecule. (- و +)
- ② water can conduct heat and electric.
- ③ water is **universal solvent** مذيب عالمي  
many substance dissolve in water.

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mixture  
 مخاليط

Combination of 2 or more substance in which each substance retains its properties.  
 اختلاط  
 تنقظ

mixture

ظلي متجانس  
 Homogeneous mixture  
 محلول (Solution)

you cannot separate them.

- water + sugar
  - water + salt
  - water + tea
  - water + coffee
- مزيج                      مزاج

ظلي غير متجانس  
 Heterogeneous mix.  
 where components remain distinct we can separate them.

- ① Salad
- ② Suspension مادة غروانية  
 water + sand will settle  
يتروب
- ③ Colloid المعلقات

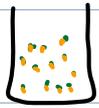
Solvent  
That dissolved  
solute.

Solute  
That dissolve in  
solvent.

particles not settle.  
- fog, milk, blood  
ink, دايونيز Paint.

Solutions

محالين



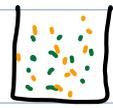
Suspensions

الغروانيات



Colloids

المعلقات



water

substance

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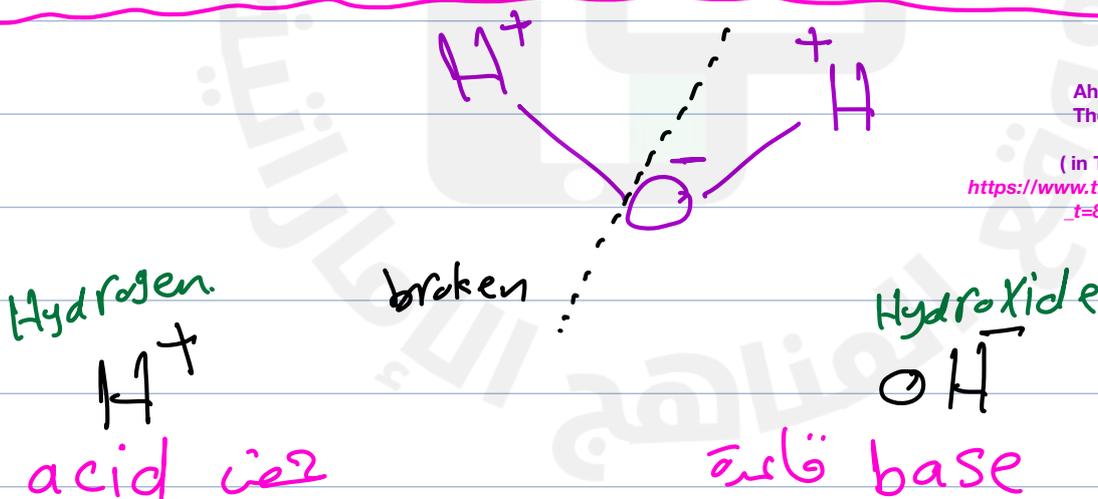
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- uniform mixture
- Homogenous
- particles don't settle.

- distinct
- Heterogenous
- particles settle over time.

- distinct
- Heterogenous
- particles don't settle.

All are mixture



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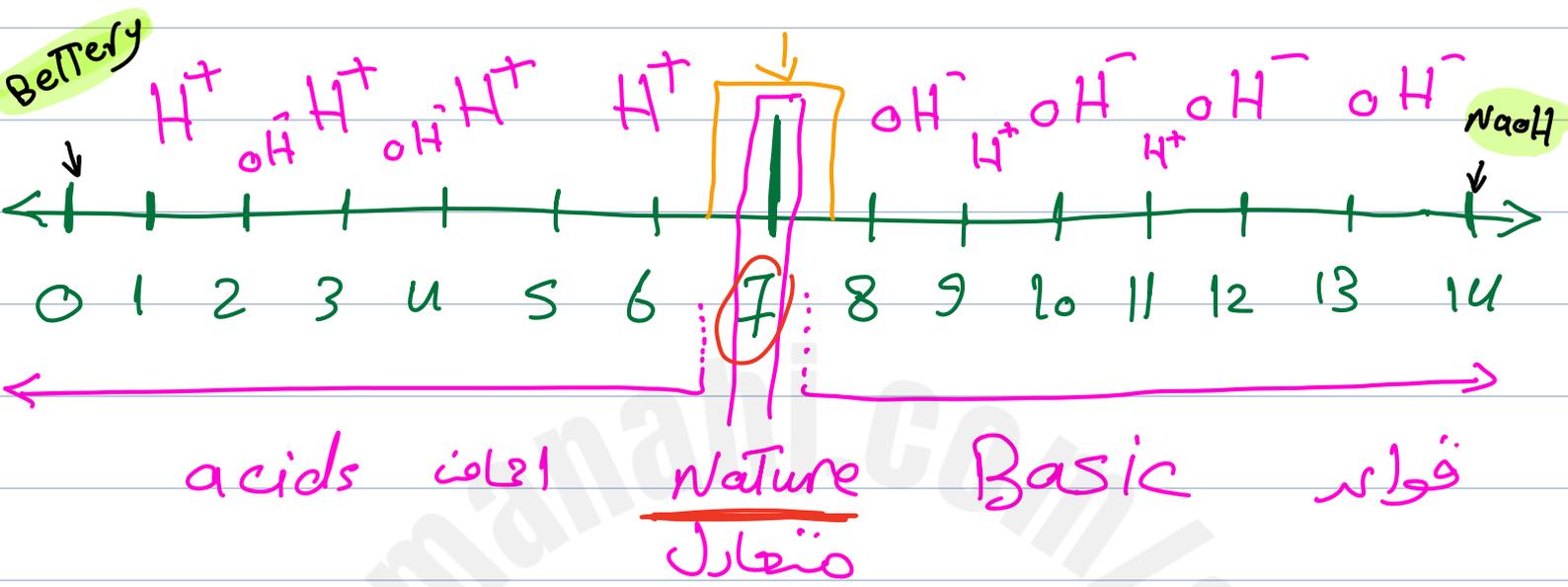
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- substance that release  $H^+$  when dissolve in water.

- substance that release  $OH^-$  when dissolve in water.

+ pH: الرقم الهيدروجيني

- Scale discover amount of  $H^+$  or  $OH^-$  in solution.
- The measure of concentration of  $H^+$  in solution.



$pH = 6$   
 $= 5$  less acid  
 $= 4$   
 $= 3$  Acid  
 $= 2$   
 $= 1$   
 $= 0$  more acid

Nature  
 Blood water

$pH = 8$  less basic  
 $= 9$   
 $= 10$   
 $= 11$  Base  
 $= 12$   
 $= 13$   
 $= 14$  more basic

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most reactions occur in  $pH = 6.5 : 7.5$

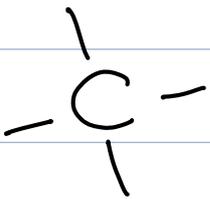
- IF pH increase or decrease will be balance when use {Buffers}

# مختلطات

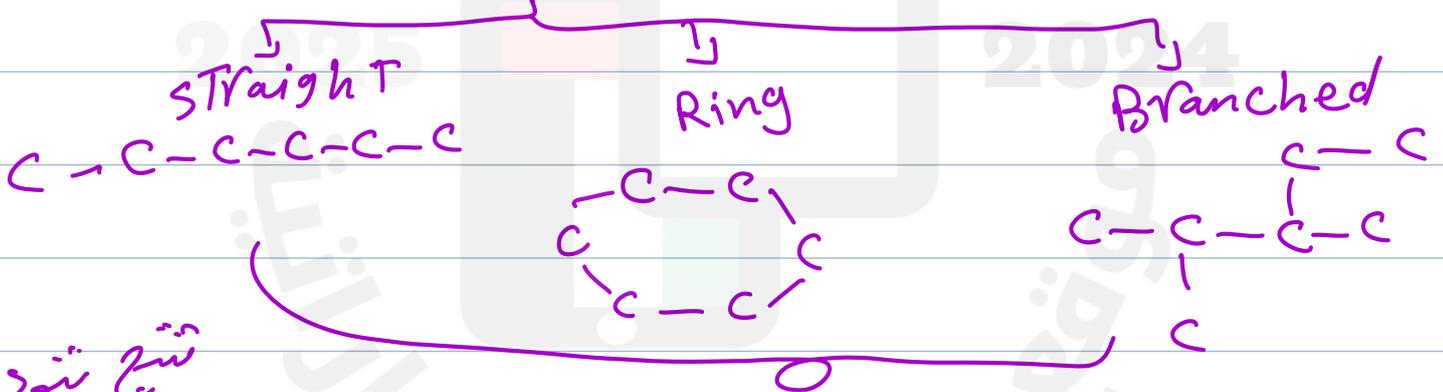
mixture can react with acid or base to keep pH within particular range.  
(6.5 - 7.5)

## The Building blocks of life الجزيئات الكبيرة (Macromolecules)

The basic element in all organic matter is [Carbon C]

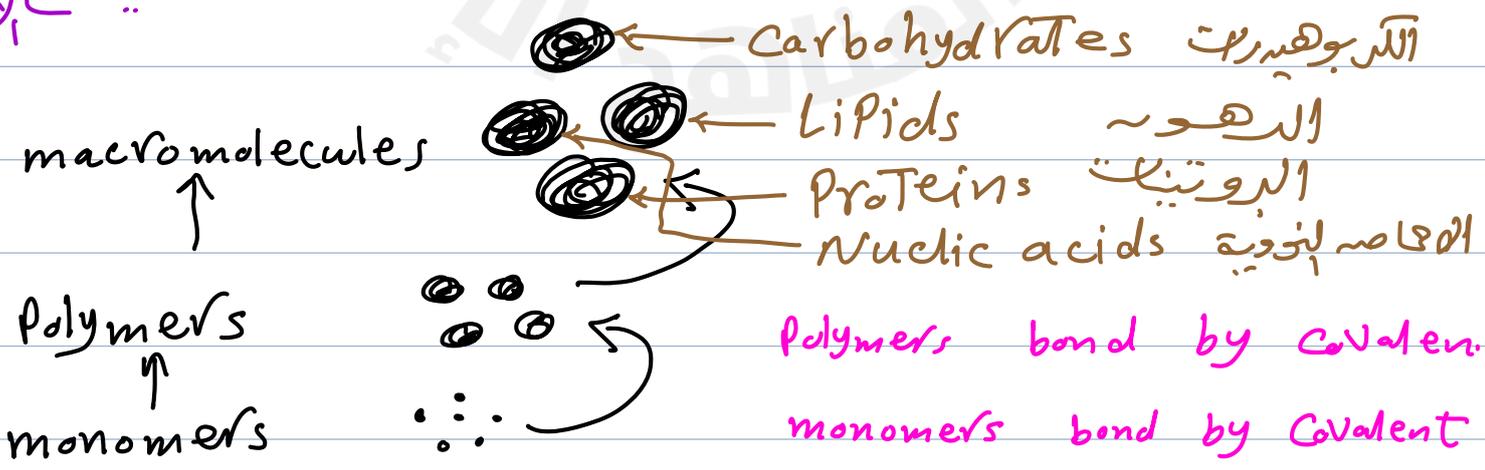


- Can form 4 covalent bond with other element.
- has 4 electrons in the outer energy level.
- Can form 3 types of chains.



تنوع تنوع  
التي تؤدي إلى تنوع الحياة على الأرض.

Resulting in diversity of life on earth.



# ① Carbohydrates C, H, O الكربوهيدرات

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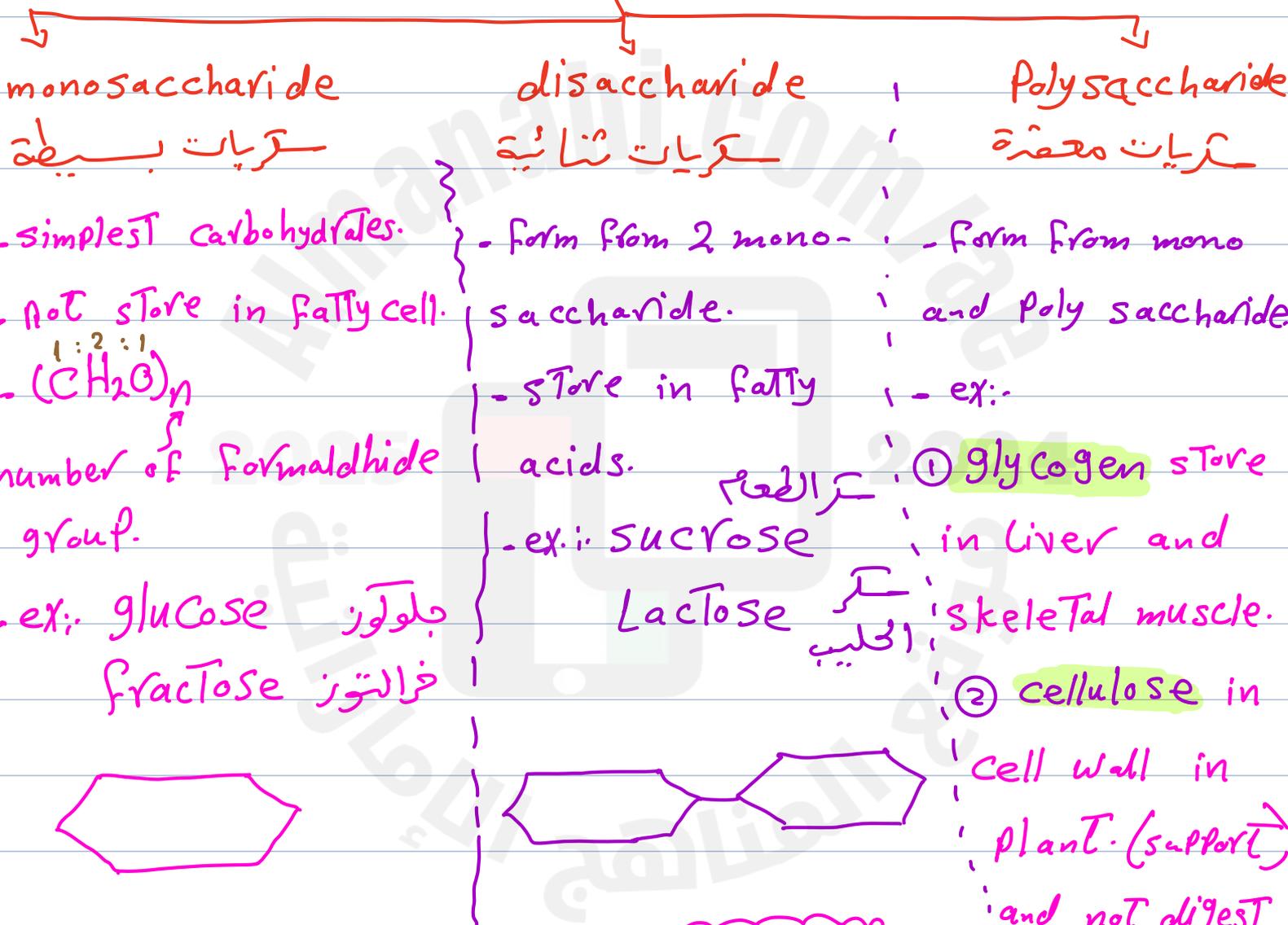
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found in Pasta, Rice, Bread, sugar, fruits.

give immediate energy. طاقة فورية

Structural support. دعم هيكل

## Carbohydrates basic unit glucose



- simplest carbohydrates.

- not store in fatty cell.



number of formaldehyde group.

- ex: glucose جلولوز

fructose فركتوز



disaccharide

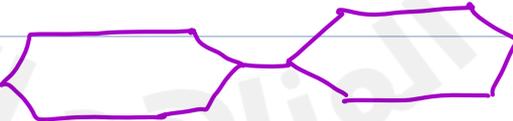
سكريات ثنائية

- form from 2 mono-saccharide.

- store in fatty acids.

- ex: sucrose سكر العسل

Lactose سكر الحليب



Polysaccharide

سكريات متعددة

- form from mono and poly saccharide.

- ex:

① glycogen store

in liver and skeletal muscle.

② cellulose in

cell wall in plant. (support)

and not digest

no enzymes can digest cellulose in bodies.

③ chitin كيتين

in insects and nails and exoskeleton of fish and fungi.

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# ② Lipids $C, H, O, P$

## الدهون

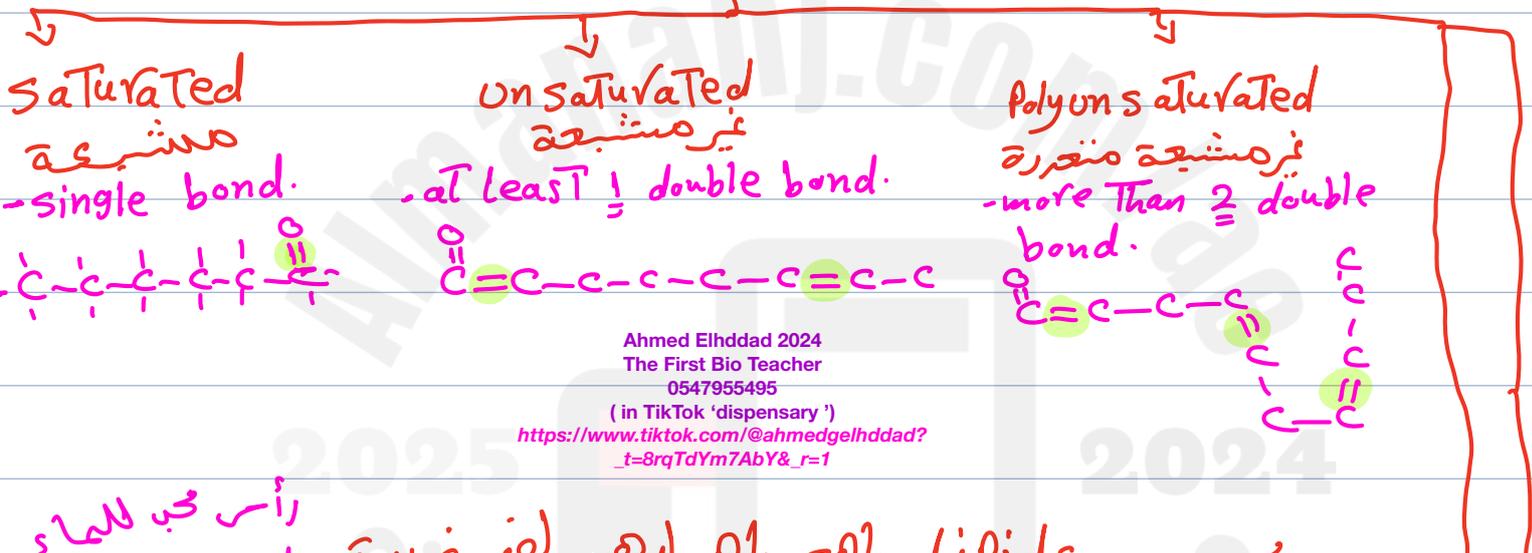
- store energy long term.
- Barriers in cells.
- not dissolve in water.

basic unit is glycerol and

3 fatty acids.

Solid (Fat)      Liquid (oil)  
 صلب                      زيت

### Lipids



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رأس محب للماء  
 hydrophilic head  
 ذيل كاره للماء  
 hydrophobic Tail

### Phospholipids دهون لفسفورية

- glycerol + 2 fatty acid + 1 phosphate group.
- found in all cell membrane.

### ستيرويدات Steroids

- cholesterol.

مفيد helpful

ضار harmful.

build vitamin D

when increase level of fat in vessels.

and sex hormones

Testosteron  
الذئرة

Estrogen  
الانوية

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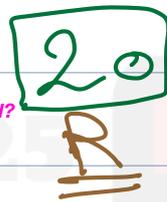
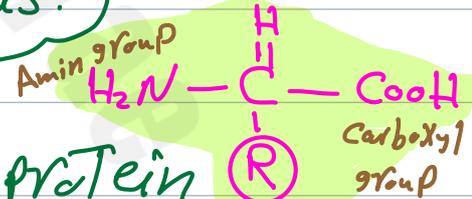
### ③ Proteins $C, H, O, N, S$ البروتينات 15% of body

- structural support.
- Transport (Hemoglobin "blood protein")
- increase speed up chemical reactions. (Enzymes).
- Cell communication.

the basic unit is

الوحدات الامينية  
Amino acids.

all amino acids have the same general structure



$\rightarrow$   $\boxed{10000}$

protein



جانب متغيرة side chain

$\boxed{15\%}$  of body weight.

- found in:

meat, poultry, Eggs, chicken, nuts, legumes.

proteins 4 levels.



each amino acid will bond with another amino acid by

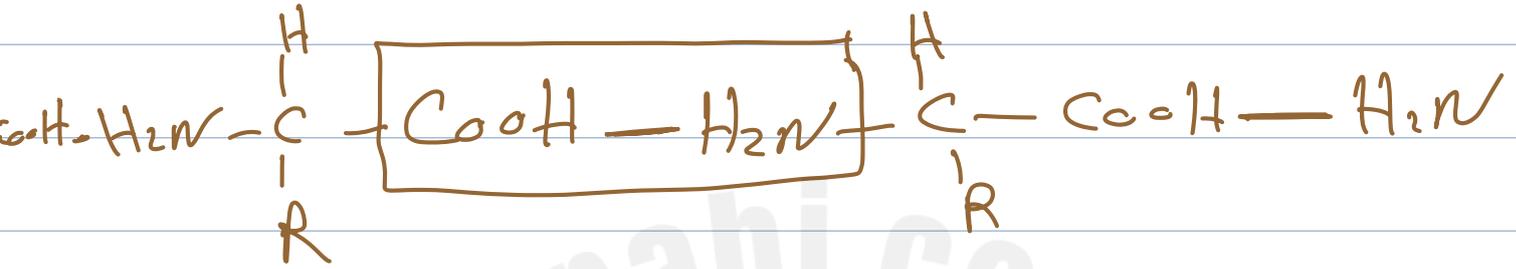
peptide bond  
رابطة ببتيدية

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Amin group + Carboxil group.  
for amino acid                      for another amino acid.



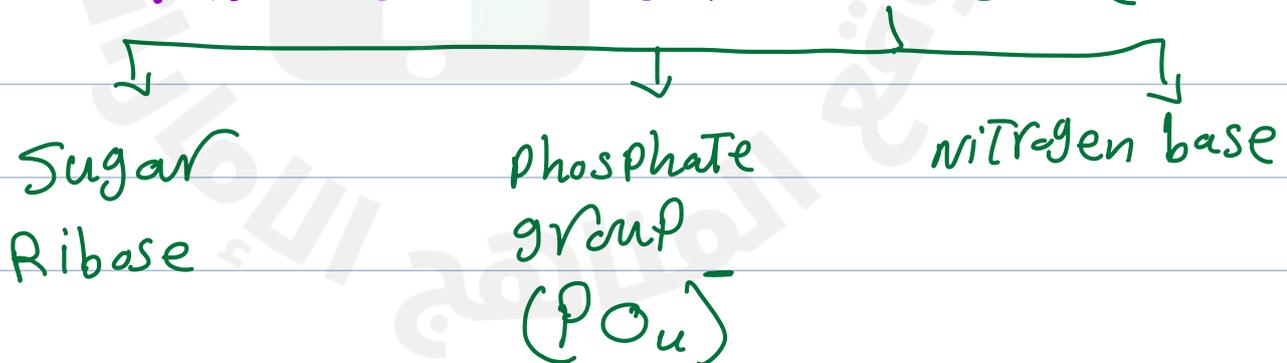
④ Nucleic acids  $C, H, O, N, P$   
الاحماض النووية P

- store and transmit genetic information.

- store chemical energy (ATP)

Adenosin triphosphate.

- Basic unit of nucleic acid is nucleotide



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# nucleic acid

## DNA



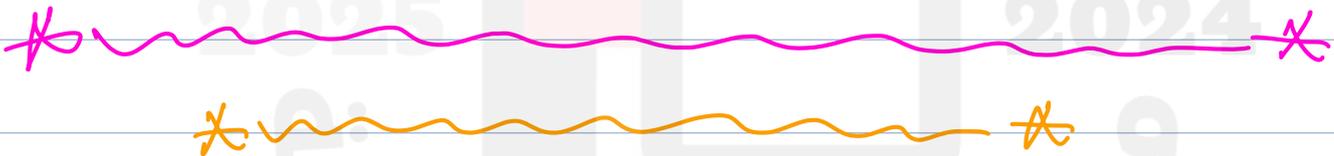
## RNA



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- double strand
- store genetic information.
- Basic in all nucleus.
- one type.

- single strand
- Copy genetic information.
- Build on DNA during transcription.
- 3 type.



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